

# ロングDCアークによる難分解性ガスSF<sub>6</sub>の分解機構 Decomposition Mechanism of SF<sub>6</sub> by Long DC Arc Plasma

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## 1. Introduction

Abatement processing with thermal plasmas has been implemented in industrial fields due to their advantages of high temperature, high chemical activity, and rapid treatment. Long DC arc has a long electrode gap distance, 300 mm. This configuration leads to sufficiently long residence time for decomposition of harmful target.

Sulfur hexafluoride, SF<sub>6</sub>, is mainly used as the insulating medium in gas circuit breaker due to its thermally and chemically stable and as a dry etching gas in the semiconductor industry. SF<sub>6</sub> has a high global warming potential about 24,000, and is difficult to be decomposed. The purpose of this study is to decompose SF<sub>6</sub> by long DC arc and to investigate decomposition mechanism.

## 2. Experiment

The setup consists of a power supply, a plasma torch, and a scrubber. The arc current was 10 A. Nitrogen at 25 L/min was used as the plasma gas, while SF<sub>6</sub> was injected at 0.5 L/min. Steam (0, 0.75, 1.5, 3.0 L/min) was introduced as the additive gas. This is because H and OH radicals inhibit recombination of SF<sub>6</sub> in the plasma region. Molar ratio of H/F was changed to 0.0, 0.5, 1.0 and 2.0.

The produced gases are analyzed by a gas chromatograph (GC) and a quadrupole mass spectrometer (QMS) to investigate the destruction and removal efficiency (DRE) and the composition of the produced gases.

## 3. Results and Discussion

**Figure 1** shows the relationship between the H/F molar ratio and DRE. The DRE is 0% at H/F=0.0, because S and F radicals were recombined to SF<sub>6</sub> after thermal decomposition. The DRE value increased as the H/F ratio. The maximum DRE value was 97% at H/F=2.0. This is because F and H combine to form stable HF after decomposition of SF<sub>6</sub> and H<sub>2</sub>O.

The gas component after SF<sub>6</sub> decomposition were analyzed by QMS. Obtained mass spectra were shown in **Fig. 2** SF<sub>6</sub> and SO<sub>2</sub>F<sub>2</sub> peaks were decreased

as H/F ratio was increased. SO<sub>2</sub>F<sub>2</sub> was derived from oxidation of SF<sub>x</sub> intermediates. Fluorine necessary for SF<sub>x</sub> recombination was completely recovered to form HF. This resulted in undetectable SO<sub>2</sub>F<sub>2</sub> at high H/F ratio.

## 4. Conclusion

SF<sub>6</sub> was successfully decomposed by long DC arc with steam addition. Long DC arc system is expected to play an active role in the semiconductor industry due to the ability to decompose alternative PFC gases completely.

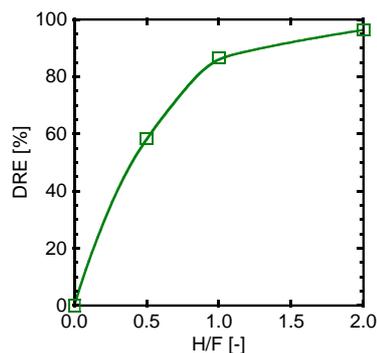


Fig. 1 Effect of H/F ratio on the DRE, SF<sub>6</sub> flow rate: 0.5 L/min, H<sub>2</sub>O flow rate: 0.0, 0.75, 1.5, 3.0 L/min.

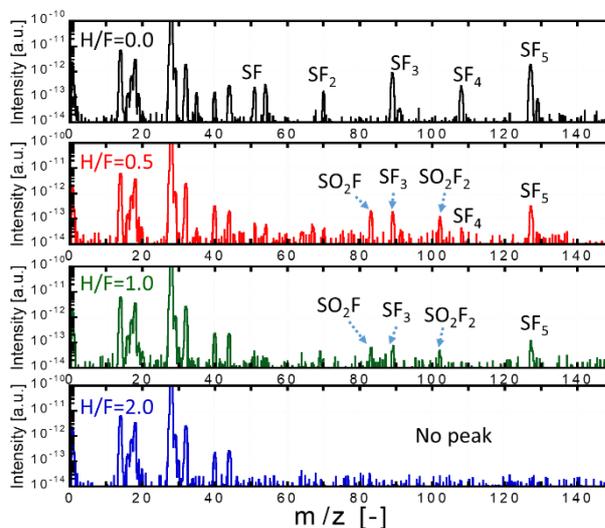


Fig. 2 Spectra of produced gas at various H/F ratio; H/F=0.0, 0.5, 1.0, 2.0.